

Statistical Physics Qualifying Exam (2025)

1. Assuming that the entropy S and the statistical number Ω of a physical system are related through an arbitrary functional form

$$S = f(\Omega),$$

show that the additive character of S and the multiplicative character of Ω *necessarily* require that the function $f(\Omega)$ be of the form $S = k \ln \Omega$. [25 points]

2. In the thermodynamic limit (when the extensive properties of the system become infinitely large, while the intensive ones remain constant), the q -potential of the system may be calculated by taking only the largest term in the sum

$$\sum_{N_r=0}^{\infty} z^{N_r} Q_{N_r}(V, T).$$

Verify this statement and interpret the result physically. [25 points]

3. Show that the entropy of an ideal gas in thermal equilibrium is given by the formula

$$S = k \sum_{\epsilon} [\langle n_{\epsilon} + 1 \rangle \ln \langle n_{\epsilon} + 1 \rangle - \langle n_{\epsilon} \rangle \ln \langle n_{\epsilon} \rangle]$$

in the case of *bosons* and by the formula

$$S = k \sum_{\epsilon} [-\langle 1 - n_{\epsilon} \rangle \ln \langle 1 - n_{\epsilon} \rangle - \langle n_{\epsilon} \rangle \ln \langle n_{\epsilon} \rangle]$$

in the case of *fermions*. [25 points]

4. Deduce the virial expansion

$$\frac{PV}{NkT} = \sum_{l=1}^{\infty} a_l \left(\frac{\lambda^3}{v} \right)^{l-1}$$

from

$$\frac{P}{kT} = -\frac{2\pi(2mkT)^{3/2}}{h^3} \int_0^{\infty} x^{1/2} \ln(1 - ze^{-x}) dx = \frac{1}{\lambda^3} g_{5/2}(z)$$

and

$$\frac{N - N_0}{V} = \frac{2\pi(2mkT)^{3/2}}{h^3} \int_0^{\infty} \frac{x^{1/2} dx}{z^{-1}e^x - 1} = \frac{1}{\lambda^3} g_{3/2}(z),$$

and obtain the following virial coefficients: a_1 , a_2 , a_3 , and a_4 . [25 points]